



‘সমানো মন্ত্র: সমিতি: সমানী’

UNIVERSITY OF NORTH BENGAL

B.Sc. Honours 3rd Semester Examination, 2022

CC6-CHEMISTRY**ORGANIC CHEMISTRY-II**

Time Allotted: 2 Hours

Full Marks: 40

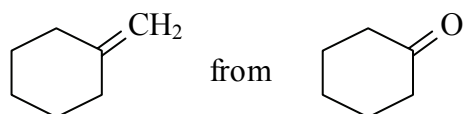
*The figures in the margin indicate full marks.***GROUP-A**

1. Answer any **five** questions from the following: 1×5 = 5
- (a) Which of the following S_N1 reactions would you expect to take place more rapidly and why?
- (i) $(\text{CH}_3)_3\text{C}-\text{I} + \text{CH}_3\text{OH} \rightarrow (\text{CH}_3)_3\text{COCH}_3 + \text{HI}$
- (ii) $(\text{CH}_3)_3\text{C}-\text{Cl} + \text{CH}_3\text{OH} \rightarrow (\text{CH}_3)_3\text{COCH}_3 + \text{HCl}$
- (b) Pyridine unlike benzene undergo faster aromatic nucleophilic substitution reaction. Why?
- (c) Halocyclopropanes are unreactive towards S_N2 pathway. Why?
- (d) Which solvent do you prefer to work with LAH?
- (e) Aromatic hydrocarbons are often identified using picric acid. Why?
- (f) Why cyanide ion is regarded as a unique reagent for benzoin condensation?
- (g) Why does the dehydration of $\text{CH}_3\text{CH}(\text{OH})\text{CH}_2\text{CHO}$ occur readily in presence of base on heating?
- (h) Explain the reaction of pyrrole with methyl magnesium iodide.

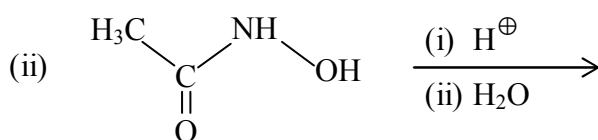
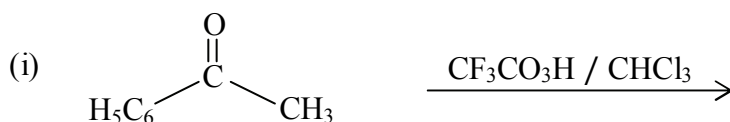
GROUP-B

2. Answer any **three** questions from the following: 5×3 = 15
- (a) (i) Explain why the hydrolysis of methyl chloride with aqueous NaOH is essentially irreversible. 2½×2 = 5
- (ii) Predict the product of the following reaction:
- $$\begin{array}{c} \text{Ph} \\ \diagdown \\ \text{C} = \text{N} - \text{OH} \\ \diagup \\ p\text{CH}_3\text{C}_6\text{H}_4 \end{array} \xrightarrow[2. \text{H}_2\text{O}]{1. \text{PCl}_5 / \text{Et}_2\text{O}} ?$$
- (b) (i) With the help of examples how can you justify the higher reactivity of organolithium than the Grignard reagent? 3
- (ii) $\text{EtSCH}_2\text{CH}_2\text{Cl}$ undergoes hydrolysis very much faster than $\text{EtOCH}_2\text{CH}_2\text{Cl}$. Explain. 2

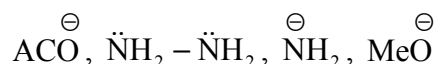
- (c) (i) What are ylides? Why are sulphur ylides more stable than nitrogen ylides? 2
 (ii) How can you differentiate between 3-pentanone and 2-pentanone by a chemical test? $1\frac{1}{2}$
 (iii) Show the scheme of preparation of $1\frac{1}{2}$



- (d) Predict the products with suitable mechanism: $2\frac{1}{2} \times 2 = 5$



- (e) (i) Arrange the following nucleophiles with increasing order of nucleophilicity: $2\frac{1}{2}$



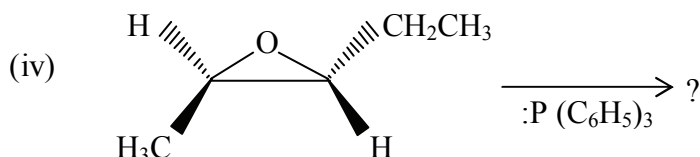
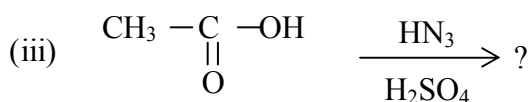
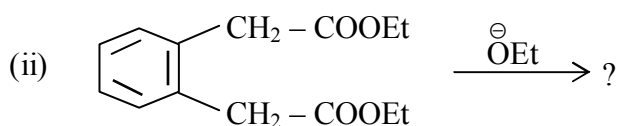
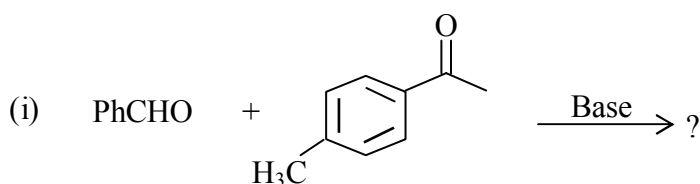
Justify your order of choice.

- (ii) Explain why benzene diazonium chloride couples with N, N-dimethyl aniline $2\frac{1}{2}$ but not with its 2, 6-dimethyl derivative.

GROUP-C

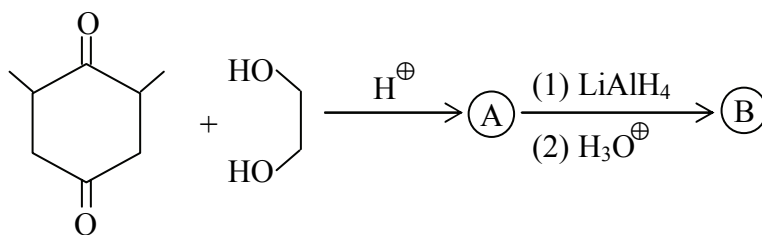
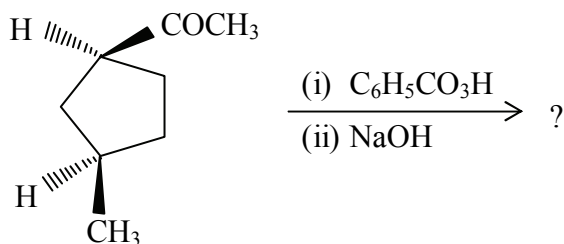
3. Answer any *two* questions from the following: $10 \times 2 = 20$

- (a) Predict the product(s) and suggest plausible mechanism: $2\frac{1}{2} \times 4 = 10$



- (b) (i) Compare the effectiveness of acetate, phenoxide and benzene sulfonate anions as leaving group if the strength of their conjugate acids are given by their pK_a values of 4.5, 10.0 and 2.6 respectively. 2

- (ii) Carry out the following transformations with mechanism: $2\frac{1}{2} \times 2 = 5$



- (iii) What happens when benzil undergoes base-catalysed rearrangement reaction in presence of MeONa instead of NaOH? 3
- (c) (i) How does CH_3COCl and $\text{CH}_3\text{CH}=\text{O}$ differ when reacted with HCN. $2\frac{1}{2} \times 4 = 10$
- (ii) How would you proceed to synthesize $\text{CH}_3\text{CH}=\text{CHCOOH}$ from CH_3COOH using a Grignard reagent as an intermediate?
- (iii) Explain why acetamide cannot be used for acetylating organic compounds.
- (iv) Convert ethylene into succinic acid.
- (d) (i) Starting from suitable substrates give a synthesis of dimedone. Give mechanism for the steps involved. 7
- (ii) Starting from acetaldehyde suggest a synthetic route to Pentaerythritol. 3

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