

UNIVERSITY OF NORTH BENGAL B.Sc. Honours 4th Semester Examination, 2022

CC9-CHEMISTRY

ORGANIC CHEMISTRY

Full Marks: 40

 $1 \times 5 = 5$

 $5 \times 3 = 15$

2+3 = 5

The figures in the margin indicate full marks. All symbols are of usual significance.

- 1. Answer any *five* questions of the following:
 - (a) Draw the structures of nicotine and reserpine.
 - (b) What is isoprene rule?
 - (c) Why only secondary amines are preferred for Mannich Reaction?
 - (d) Furan and pyrrole have opposite direction of dipole moment Why?
 - (e) Cyclohexylamine is more basic than aniline Why?
 - (f) Trans decalin is more stable than cis-decalin. Why?
 - (g) Chalk out a synthetic plan for indene.

2. Answer any *three* questions from the following:

(a) (i) Arrange the following in decreasing order of their reactivity toward 3+2=5 electrophilic substitution reaction. Give a suitable explanation in favor of your choice.

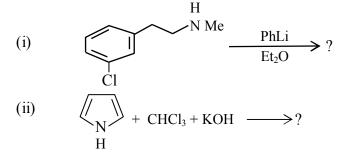
Benzene; thiophene and pyrrole.

- (ii) Why the Gabriel phthalimide synthesis of amine is not suitable for the preparation of tertiary amine?
- (b) (i) Write the structures of different stereoisomers of menthol.
 - (ii) In the following reaction, which one will be major product and why?

$$\xrightarrow{CH_3} E^{\bigoplus} \xrightarrow{E} CH_3 \xrightarrow{CH_3} E^{\bigoplus}$$

(c) What would be the major product in the given reactions? Draw the plausible $2\frac{1}{2} \times 2 = 5$ reaction mechanism in each case.

 \mathbf{D}



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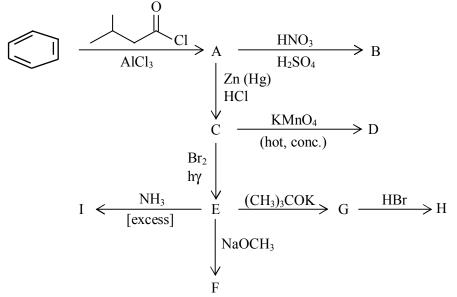
- (d) (i) How do you prove that nicotine has $N CH_3$ group? 2+3 = 5
 - (ii) How do you establish the positions of unsaturation in citral?
- (e) (i) *N*-methyl-2-pyrrole carboxaldehyde does not undergo Cannizzaro reaction. 2+3 = 5— Why?

5 + 5 = 10

- (ii) Unlike benzene, pyridine undergoes both electrophilic and nucleophilic substitution reaction. Why? Identify the position(s) of pyridine ring where substitution will occur in each case.
- 3. Answer any *two* questions from the following: $10 \times 2 = 20$
 - (a) (i) An organic compound, with molecular formula $C_8H_{17}N$, which upon treatment with excess MeI followed by silver oxide and heating, gives pure (S)-enantiomer of N,N-dimethyl oct-7-ene-4-amine. Propose a complete structure of the unknown compound, and show how this reaction gives the observed product.
 - (ii) Devise a synthesis for each of the following, starting with any compounds containing no more than six carbon atoms.

(A)
$$CH_3 - CH_2 - O$$
 H H H O
(B) HO $CH_2 - CH - CH_3$
 $NH - CH_3$
(C) HO HO $CH_2 - CH_2 - NH_2$

(b) (i) Give the structures of compounds A to I in the following series of 5+5=10 reactions.



(ii) In the following reaction, identify both X and Y. Propose a mechanism for the synthesis of X. What kind of structural change occurs under basic medium? Also propose a plausible mechanism.

Phthalic anhydride + 2.Phenol
$$\xrightarrow{H_2SO_4}$$
 X \xleftarrow{OH} Y

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(c) Predict the product with suitable mechanism.

(i)
$$\underbrace{N_{H}}_{H} \underbrace{MeI}_{DMF,}?$$
(ii)
$$\underbrace{MeI}_{DMF,}?$$
(i)
$$\underbrace{MeI}_{DMF,}?$$
(i) \\ \underbrace{MeI}_{DMF,}?
(i)
$$\underbrace{MeI}_{DMF,}?$$
(i) \\ \underbrace{MeI}_{

(ii)
$$\langle N \rangle$$
 + HCHO + Me₂NH $\xrightarrow{1. \text{ H}}$?
H

(iii)
$$\langle \langle S \rangle + C_6H_5COCl \xrightarrow{SnCl_4} ?$$

(iv)
$$H_2SO_4 \rightarrow P$$

__×____

 $2 \times 5 = 10$